

Chirality Transfer during the [2,3]-sila-Wittig Rearrangement and Cyclopropanation Reaction of Optically Active [(*sec*-Allyloxy)silyl]lithiums

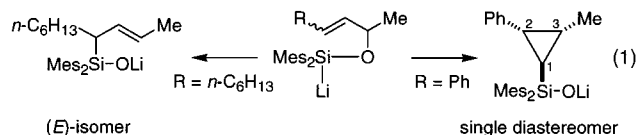
Atsushi Kawachi,* Hirofumi Maeda, Hiroshi Nakamura, Noriyuki Doi, and Kohei Tamao*

Institute for Chemical Research
Kyoto University, Uji, Kyoto 611-0011, Japan

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Intramolecular reactions of organolithium reagents with olefins have been extensively studied in organic synthesis. One representative reaction is the Wittig rearrangement¹ of α -alkoxy-organolithium compounds, which provides a versatile method for the regio- and stereoselective C–C bond formation along with the allylic transposition. Another topic is the intramolecular carbolithiation of olefins,² which allows regio- and stereoselective cyclizations. Much attention has recently been paid to asymmetric variants of these reactions.^{1b–e,2a–d}

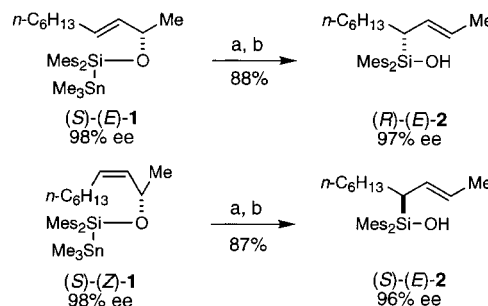
In contrast to this, less attention has been paid to the intramolecular reactions of silyllithium reagents³ with olefins despite the potential utility for the regio- and stereoselective Si–C bond formations. Recently, we found two reaction modes for the intramolecular rearrangements of the [(*sec*-allyloxy)dimesitylsilyl]lithiums,⁴ as shown in eq 1. One is the [2,3]-sila-Wittig rear-



angement, the silicon analogues to the [2,3]-Wittig rearrangement,⁵ that is, the [(allyloxy)silyl]lithium bearing an alkyl group on the terminus of the olefin undergoes the [2,3]-rearrangement to afford the (*E*)-allylsilanes. The other is the cyclopropanation reaction in which the [(allyloxy)silyl]lithium bearing a phenyl group on the terminus of the olefin gives the corresponding substituted cyclopropylsilyl ether as a single diastereoisomer.⁶ We now disclose the chirality transfer of these reactions using enantio-enriched *sec*-allylic alcohol derivatives; the center of chirality at one allylic carbon atom is intramolecularly transferred to the newly formed stereogenic centers, which leads to optically active allylsilanes⁷ and cyclopropylsilyl ether.^{8,9}

We first investigated the 1,3-chirality transfer during the [2,3]-sila-Wittig rearrangement, as shown in Scheme 1. Treatment of the [(*sec*-allyloxy)dimesitylsilyl]stannane¹⁰ (*S*)-(*E*)-**1** (98% ee) with *n*-BuLi (2.0 mol amt.) in THF at 0 °C for 3 h provided the

Scheme 1^a



^a (a) *n*-BuLi (×2), THF, 0 °C, 3 h. (b) 5% NH₄Cl aq.

allylsilane (*R*)-(*E*)-**2** in 88% yield with 97% ee.¹¹ In contrast, (*S*)-(*Z*)-**1** (98% ee) afforded the allylsilane (*S*)-(*E*)-**2** in 87% yield with 96% ee. Thus, the resulting stereochemistry of the new chiral center significantly depends on the olefin geometry of the substrate, while the resulting olefin geometry is always *E*.

These aspects provide an insight into the mechanism,¹² as shown in Scheme 2. For instance, the silyllithium (*S*)-(*E*)-**3** derived from (*S*)-(*E*)-**1** undergoes suprafacial attack on the olefin in the lower-energy conformer (*S*)-(*E*)-**3a**, to give the observed (*R*)-(*E*)-**2**, whereas the higher energy conformer (*S*)-(*E*)-**3b** due to the allylic strain would provide (*S*)-(*Z*)-**2**, but this is not experimentally observed. This is consistent with the general attributes of the [2,3]-Wittig rearrangement.^{1a}

(7) Synthesis of optically active allylsilanes: (a) For a review; Sarkar, T. K. *Synthesis* **1990**, 969 and 1101. (b) Catalytic asymmetric cross-coupling of α -(silyl)alkyl Grignard reagents with alkenyl bromides; Hayashi, T.; Konishi, M.; Okamoto, Y.; Kabeta, K.; Kumada, M. L. *J. Org. Chem.* **1986**, *51*, 3772. (c) Claisen rearrangement of optically active allylic alcohol derivatives; Sparks, M. A.; Panek, J. S. *J. Org. Chem.* **1991**, *56*, 3431. (d) Nucleophilic substitution of optically active allyl esters and carbamates with (organosilyl)cuprates; Fleming, I.; Higgins, D.; Lawrence, N. J.; Thomas, A. P.; *J. Chem. Soc., Perkin Trans. 1* **1992**, 3331. (e) Wittig olefination of optically active α -silylaldehydes; Bhushan, V.; Lohray, B. B.; Enders, D. *Tetrahedron Lett.* **1993**, *34*, 5067. (f) Catalytic asymmetric hydrosilylation of 1,3-dienes; Hayashi, T.; Matsumoto, Y.; Morikawa, I.; Ito, Y. *Tetrahedron Asymmetry* **1990**, *1*, 151. (g) Hatanaka, Y.; Goda, K.; Yamashita, F.; Hiyama, T. *Tetrahedron Lett.* **1994**, *35*, 7981. (h) Catalytic asymmetric silylation of allylic chlorides with chlorodisilanes; Hayashi, T.; Ohno, A.; Lu, S.-J.; Matsumoto, Y.; Fukuyo, E.; Yanagi, K.; *J. Am. Chem. Soc.* **1994**, *116*, 4221. (i) Palladium-catalyzed bis-silylation of optically active allylic alcohols; Sugimoto, M.; Matsumoto, A.; Ito, Y. *J. Am. Chem. Soc.* **1996**, *118*, 3061. (j) Substitution reactions of alkylmagnesium reagents with optically active silyl-substituted allylic carbamates; Smitrovich, J. H.; Woerpel, K. A. *J. Org. Chem.* **2000**, *65*, 1601.

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(9) Other methods for synthesis of optically active cyclopropanes, e.g., for a general review: (a) Salaün, J. *Chem. Rev.* **1989**, *89*, 1247. Asymmetric variants of Simmons–Smith reaction: (b) Charette, A. B.; Marcoux, J.-F. *Synlett* **1995**, 1197 and references therein. (c) Denmark, S. E.; O'Connor, S. P. *J. Org. Chem.* **1997**, *62*, 584. Transition metal-catalyzed asymmetric cyclopropanation of olefins with diazoacetates: (d) Pfaltz, A. *Acc. Chem. Res.* **1993**, *26*, 339 and references therein. (e) Park, S.-B.; Sakata, N.; Nishiyama, H. *Chem. Eur. J.* **1996**, *2*, 303.

(10) The [(allyloxy)dimesitylsilyl]stannanes were prepared by the reaction of the corresponding enantio-enriched *sec*-allylic alcohols and the chloro-(dimesityl)silylstannane in THF in the presence of triethylamine and 4-(dimethylamino)pyridine; see ref 5. The optically active allylic alcohols were derived from commercially available (*S*)-(-)-3-buten-2-ol (98% ee); see Supporting Information.

(11) The enantiomeric excess of **2** was directly determined by chiral HPLC analysis using CHIRALPAK AD column eluted with hexane/2-propanol (50/1). The absolute configuration of the allylsilanes was determined after conversion to the allylic alcohols by *m*-chloroperbenzoic acid according to the literature; Hayashi, T.; Okamoto, Y.; Kabeta, K.; Hagihara, T.; Kumada, M. *J. Org. Chem.* **1984**, *49*, 4224, and see ref 7i.

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(2) Intramolecular carbolithiation, e.g.: (a) Coldham, I.; Hufton, R.; Snowden, D. J. *J. Am. Chem. Soc.* **1996**, *118*, 5322. (b) Woltering, M. J.; Fröhlich, R.; Hoppe, D. *Angew. Chem., Int. Ed. Engl.* **1997**, *36*, 1764. (c) Hoppe, D.; Woltering, M. J.; Oestreich, M.; Fröhlich, R. *Helv. Chim. Acta* **1999**, *82*, 1860. (d) Tomooka, K.; Komine, N.; Sasaki, T.; Shimizu, H.; Nakai, T. *Tetrahedron Lett.* **1998**, *39*, 9715. (e) Cheng, D.; Knox, K. R.; Cohen, T. *J. Am. Chem. Soc.* **2000**, *122*, 412.

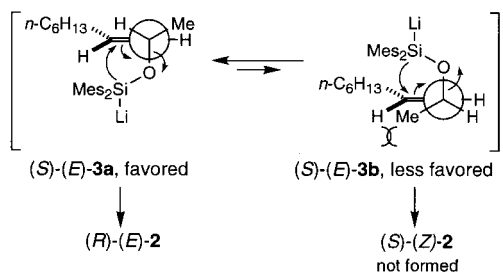
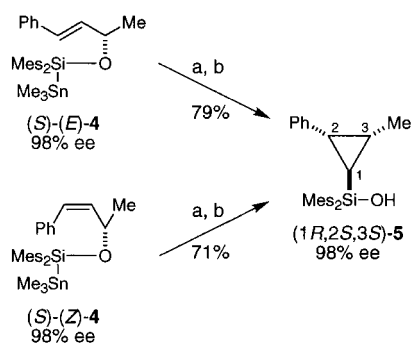
(3) For reviews of silyl anions: (a) Tamao, K.; Kawachi, A. *Adv. Organomet. Chem.* **1995**, *38*, 1. (b) Lickiss, P. D.; Smith, C. M. *Coord. Chem. Rev.* **1995**, *145*, 75. (c) Belzner, J.; Dehnert, U. In *The Chemistry of Organic Silicon Compounds*; Rappoport, Z.; Apeloig, Y., Eds.; John-Wiley & Sons: Chichester, 1998; Vol. 2, pp 779–826.

(4) The term “*sec*-allylic” means that the allylic carbon is secondary.

(5) Kawachi, A.; Doi, N.; Tamao, K. *J. Am. Chem. Soc.* **1997**, *119*, 233.

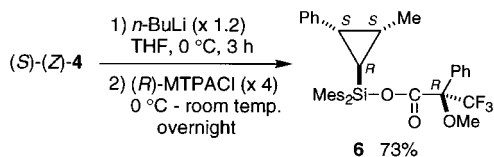
(6) Kawachi, A.; Maeda, H.; Tamao, K. *Chem. Lett.* **2000**, 1216.

Scheme 2

Scheme 3^a

^a (a) *n*-BuLi (×1.2), THF, 0 °C, 3 h. (b) 5% NH₄Cl aq.

Scheme 4

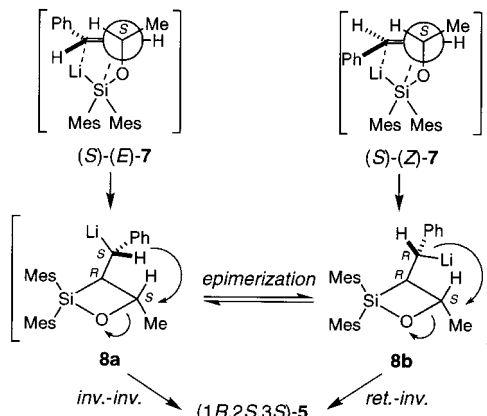


We next applied this chirality transfer technique to the cyclopropanation reaction, the other reaction mode of the [(*sec*-allyloxy)silyl]lithiums,⁶ in which the sense of chirality at the allylic position in the substrate would define the absolute configuration of three stereocenters in the resulting cyclopropane. The results are shown in Scheme 3. Thus, treatment of the enantio-enriched (S)-(E)-4 (98% ee)¹⁰ with *n*-BuLi (1.2 mol amt.) in THF at 0 °C for 3 h afforded (1*R*,2*S*,3*S*)-5 in 79% yield with 98% ee.¹³ Under the same reaction conditions, (S)-(Z)-4 (98% ee) gave the same stereoisomer (1*R*,2*S*,3*S*)-5 in 71% yield with 98% ee. The absolute configuration (1*R*,2*S*,3*S*) of 5 was definitely determined by the X-ray crystallographic analysis of the ester 6 obtained by treatment with (R)- α -methoxy- α -(trifluoromethyl)phenylacetyl chloride, as shown in Scheme 4. Thus, the resulting stereochemistry depends on the absolute configuration of the allylic carbon independent of the olefin geometry in 4, in sharp contrast to the [2,3]-sila-Wittig rearrangement mentioned above.

The stereochemical course is rationalized as follows.⁶ An intramolecular *syn*-lithiosilylation via 7 proceeds in the initial step, facilitated by the stabilizing phenyl residue, as shown in Scheme 5. The intermediate 8a is formed from (S)-(E)-4 via (S)-(E)-7,

(13) The enantiomeric excess of 5 was directly determined by chiral HPLC analysis using CHIRALCEL OD column eluted with hexane/2-propanol (100/1.5).

Scheme 5



whereas 8b results from (S)-(Z)-4 via (S)-(Z)-7. The epimers 8a and 8b are expected to be configurationally labile at the benzylic center and are in equilibrium.^{2b,c} Subsequently, cyclopropanation of the oxasilacyclobutanes 8a and 8b takes place from a conformation, avoiding steric strain between the phenyl residue and the mesityl group(s), with inversion at the C–O bond. For the formation of (1*R*,2*S*,3*S*)-5 from 8a, inversion of the benzylic center is required, whereas starting from 8b, retention leads to the observed stereochemistry.¹⁴

In conclusion, we have achieved the chirality transfer in the [2,3]-sila-Wittig rearrangement and the cyclopropanation reaction of the chiral [(*sec*-allyloxy)silyl]lithiums. While the former reaction allows the stereoselective formation of one Si–C bond, the latter reaction allows the stereoselective formation of one C–C bond in addition to one Si–C bond and can define the absolute configurations of three stereocenters in the cyclopropane ring. Since a variety of optically active allylic alcohols are now available by established methods,^{15,16} the present method will widen its variation. The scope and limitations are now under investigation.

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Supporting Information Available: Experimental details and X-ray structural information of 6 (PDF). An X-ray crystallographic file (CIF). This material is available free of charge via the Internet at <http://pubs.acs.org>.

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